بسم الله الرحمن الرحيم

Republic of Iraq

The Ministry Of Higher Education

& Scientific Research



University: Baghdad College: College Science for women Department: Chemistry Stage:4th Lecturer name: Dr. Saadiyah Ahmed Dhahir Qualification: PhD in Analytical chemistry Place of work: Baghdad

## Syllabus Form

Instructor Name	Dr. Saadiyah Ahmed Dhahir					
E-mail	sadiataher@yahoo.com sadiataher@csw.uobaghdad.edu.iq					
Course Title	Instrumental analysis and spectroscopy					
Course Coordinator						
Course Objectives	<ul> <li>Demonstrate knowledge of sampling methods for all states of matter.</li> <li>Recognize interferences in instrumental analysis.</li> <li>Comprehend the concept of and perform instrument and method calibration.</li> <li>Apply and assess concept of availability and evaluation of analytical standard and formulate standardization methodology.</li> <li>Integrate a fundamental understanding of the underlining physics principles as they relate to specific instrumentation used for electro analytical methods, atomic and molecular, spectrometry chromatography and thermal analysis.</li> <li>Understand and be able to apply the theory and operational principles of analytical instrument.</li> <li>Distinguish between qualitative and quantitative measurements and be able to effectively compare and critically select methods for elemental and molecular analyses.</li> </ul>					
Course Description	This course describe the basic principles and the instrumental design of a variety of analytical techniques, including:, spectrochemical (molecular and atomic), chromatographical, of analysis and covers the instrument of thermal analysis and electrochemistry with basic electronics and signal-to-noise enhancement.					
Textbook	<ul> <li>Douglas A. Skoog ,James Holler, Stanly R. Crouch., "Principles of instrumental analysis" 7<sup>th</sup> Edition, 2007.</li> </ul>					
References	<ul> <li>Douglas A. Skoog ,James Holler, Stanly R. Crouch., "Principles of instrumental analysis" 7<sup>th</sup> Edition, 2007 .</li> <li>D.C. Harris, "Quantitative Chemical Analysis", 6th edition,2003.</li> <li>Understanding Chemistry, Instrumental Analysis 2008</li> <li>John Kenkel ,"Analytical Chemistry for Technicians". Third Edition,2003.</li> </ul>					
	Term Tests	Laboratory	Quizzes	Project	Final Exam	
Course Assessments	As(26%)	As(12%)	As(2%)	-	As(60%)	

General	Define and Identify appropriate instrumental methods for certain chemical analysis
Notes	and their application in for quantitative and qualitative for different chemical
	compounds

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Course Weekly Outline

University :Baghdad College: : College Science for women Department: Chemistry Stage:4<sup>th</sup> Lecturer name: Dr. Saadiyah Ahmed Dhahir Qualification: PhD in Analytical chemistry

Place of work: Baghdad

Week	Date	Topes Covered	Lab. Experiment Assignments	Notes
1				
2				
2				
3				
4				
5				
6				
7				
8				
9				
10				
11				
12				
13				
14				
15				
16				
		Half –	year break	
17	21/2/2016	Atomic spectroscopy,	Determination of sodium and	
		Introduction	potassium by emission of flam method ,part 1	
18	28/2/2016	Atomic absorption	Determination of sodium and	
		spectroscopy, principles,	potassium by emission of flam	
		instrumentation	method ,part 2	
19	6/3/2016	Atomic emission spectroscopy,	The calibration of a combination of	
		Atomic Fluorescence	phosphoric acid and hydrochloric	
<u> </u>		spectroscopy, application	with a strong base by measurement	

			effort.		
20	13/3/2016	Inductively Coupled Plasma, principles, instrumentation, application	Infrared spectra of carboxylic acid ,phenols, aromatic amines		
21	20/3/2016	Gas chromatography, Introduction, principles	Infrared spectra of aldehydes and ketones compounds		
22	27/3/2016	Gas chromatography, instrumentation, application	spectroscopy in the visible region		
23	3/4/2016	First Exam	Colorimetric Determination of Co <sup>+2</sup>		
24	10/4/2016	Liquid chromatography, Introduction, principles	Determination of dissociation constant indicator, part 2		
25	17/4/2016	High performance liquid chromatography HPLC, Introduction, principles	Determination of dissociation constant indicator, part 1		
26	24/4/2016	High performance liquid chromatography HPLC, instrumentation, application	Spectrometric Determining of Nitrate Ion Concentration in Tap Water		
27	8/5/2016	Ion chromatography, instrumentation, application	Determination of dissociation constant indicator, part 1	1/5/2016 عطلة رسمية عيد	
28	15/5/2016	Thermal analysis, Introduction, application	Determination of dissociation constant indicator, part 2		
		Conductometry, Introduction, application	Conductometric titrations for acid strong with base weak		
29	22/5/2016	Potentiometry, Introduction, application	Oral Exam of Lab. Experiment		
30	29/5/2016	Second Exam	Theoretical Exam of Lab. Experiment		
	Instructor Signature: Dean Signature:				

Prof .Assist. Dr. Saadiyah Ahmed Dhahir

Prof. Dr. Ahlam Mohammad Farhan